

## Enantioselective Synthesis of a Chiral Intermediate for Aztreonam and Related Monobactam Antibiotics

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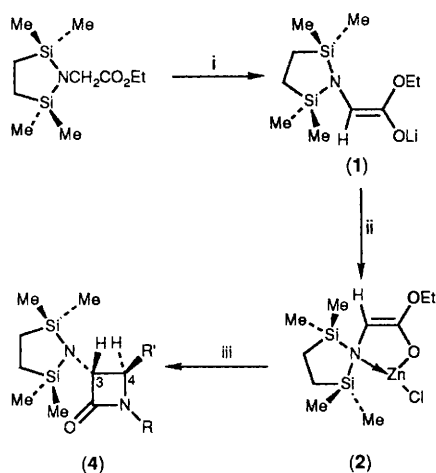
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The enantioselective synthesis of *trans*-(3*R*,4*S*)-1-(*R*)- $\alpha$ -methylbenzyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-*N*-(*R*)- $\alpha$ -methylbenzyliminoazetid-2-one has been accomplished by reaction of the chlorozinc enolate of an *N,N*-diprotected glycine ethyl ester with a chiral  $\alpha$ -diimine.

Recently, we have demonstrated that zinc enolates of disubstituted glycine esters are effective reagents for the diastereoselective synthesis of 3-amino-2-azetidones, the principal building blocks for the synthesis of aztreonam and related monobactam antibiotics.<sup>1</sup> Whereas most studies in the field of  $\beta$ -lactam synthesis have dealt with the development of enantioselective routes to chiral intermediates for thienamycin and related antibiotics,<sup>2,3</sup> little attention has been paid to the enantioselective synthesis of 3-aminoazetid-2-ones.<sup>4</sup>

Here we report the first enantioselective zinc-mediated synthesis of *trans*-(3*R*,4*S*)-1-(*R*)- $\alpha$ -methylbenzyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-*N*-(*R*)- $\alpha$ -methylbenzyliminoazetid-2-one starting from *N,N'*-bis(*R,R'*)- $\alpha$ -methylbenzyl-1,4-diazabuta-1,3-diene [(*R*)- $\alpha$ -methylbenzyl-DAB].<sup>5</sup>

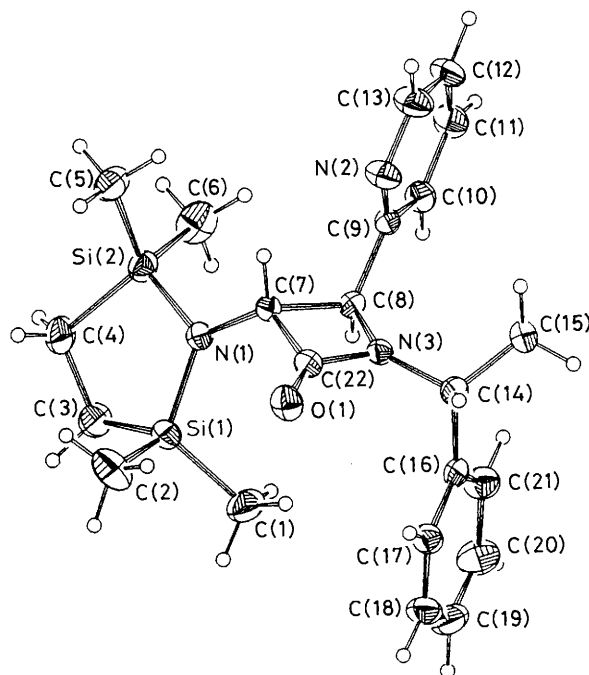


a; R = Bu<sup>t</sup>, R' = C(H)=N-R

b; R = (*R*)-C(H)(Me)Ph, R' = C(H)=N-R

c; R = (*R*)-C(H)(Me)Ph, R' = 2-pyridyl

**Scheme 1.** Reagents and conditions: i, lithium diamide (LDA) in THF, -78 °C; ii, ZnCl<sub>2</sub> in Et<sub>2</sub>O, -78 °C; iii, R-N=C(H)-R' (3), THF, -78 °C → room temperature.



**Figure 1.** An ORTEP drawing (30% probability level) of *trans*-(3*R*,4*S*)-1-(*R*)- $\alpha$ -methylbenzyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-(2-pyridyl)azetid-2-one (4c) together with the adopted numbering scheme.

We were interested in the reactions of zinc enolates with 1,4-disubstituted 1,4-diazabuta-1,3-dienes [(*R*)-DAB] to form 3-amino-4-imino- $\beta$ -lactams because the 4-imino group represents a protected aldehyde function, which is suitable for further derivatisation.<sup>6</sup> These reactions, which display a very high diastereoselectivity (Scheme 1), surprisingly proceeded also with a high enantioselectivity using the easily available (*R*)- $\alpha$ -methylbenzyl-DAB. Thus the zinc enolate (**2**) was prepared *in situ*,<sup>1</sup> and reacted with *t*-butyl-DAB (**3a**) or (*R*)- $\alpha$ -methylbenzyl-DAB (**3b**) in tetrahydrofuran (THF) at  $-70^\circ\text{C}$  to afford *trans*-1-*t*-butyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-*N*-*t*-butyliminoazetid-2-one (**4a**)<sup>†</sup> in 98% yield and *trans*-(3*R*,4*S*)-1-(*R*)- $\alpha$ -methylbenzyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-*N*-(*R*)- $\alpha$ -methylbenzyliminoazetid-2-one (**4b**)<sup>†</sup> in 90% chemical and 86% optical<sup>‡</sup> yields respectively. The latter compound is a threefold protected intermediate for the synthesis of aztreonam and related monobactam antibiotics.

Zinc enolate (**2**) also reacted smoothly with *N*-(*R*)- $\alpha$ -methylbenzyl(2-pyridyl)carbalimine (**3c**), a chiral imine containing a 1,4-diazabuta-1,3-diene skeleton, to afford *trans*-(3*R*,4*S*)-1-(*R*)- $\alpha$ -methylbenzyl-3-(2,2,5,5-tetramethyl-1-aza-2,5-disilacyclopentyl)-4-(2-pyridyl)azetid-2-one (**4c**)<sup>†</sup> in 98% chemical and  $\geq 95\%$  optical yield. The absolute configuration of (**4c**) and therefore also of (**4b**) was determined to

be 3*R*,4*S* from an *X*-ray structure determination,<sup>§</sup> the result of which is shown in Figure 1.

The reactions of zinc enolate (**2**) with (*R*)-DAB substrates proved to be superior to those of the corresponding lithium enolate (**1**) under the same reaction conditions. Starting from the latter enolate (**1**) compound (**4a**) was produced in only 15% yield, (**4b**) in 45% chemical and 40% optical yield, and (**4c**) in 70% chemical and 50% optical yield. A discussion on the difference in reactivity of lithium and zinc enolates will be presented in a forthcoming paper.

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§ *Crystal data* for (**4c**):  $\text{C}_{22}\text{H}_{31}\text{N}_3\text{OSi}_2$ ,  $M = 409.68$ , monoclinic, space group *I*2,  $a = 18.410(1)$ ,  $b = 6.813(1)$ ,  $c = 19.471(1)$  Å,  $\beta = 105.77(1)^\circ$ ,  $U = 2350.3(4)$  Å<sup>3</sup>,  $Z = 4$ ,  $F(000) = 880$ ,  $D_c = 1.158$  g cm<sup>-3</sup>,  $T = 295$  K, Mo- $K_\alpha$  (Zr-filtered) radiation ( $\lambda = 0.71073$  Å),  $\mu(\text{Mo-}K_\alpha) = 1.40$  cm<sup>-1</sup>. A redundant set of 6910 reflections was collected on an Enraf-Nonius CAD4F diffractometer in the range  $2.7 \leq 2\theta \leq 55^\circ$ . 2195 Unique reflections ( $R_{av} = 0.04$ ) with  $I \geq 2.5\sigma(I)$  were used in the structure solution (direct methods; SHELXS-86) and weighted full-matrix least-squares refinement (SHELX-76), which converged at  $R$  and  $R_w$  values of 0.046 and 0.044 respectively. The enantiomorph was chosen on the basis of the configuration of the starting (*R*)- $\alpha$ -methylbenzylamine. Atomic coordinates, bond lengths and angles, and thermal parameters have been deposited at the Cambridge Crystallographic Data Centre. See Notice to Authors, Issue No. 1.

† The new compounds (**4**) gave IR and <sup>1</sup>H and <sup>13</sup>C NMR spectra and elemental analyses consistent with the assigned structures. 200 MHz <sup>1</sup>H NMR(CDCl<sub>3</sub>) (**4a**):  $\delta$  7.46 (d, 1H,  $J$  7.8 Hz, HC=N), 4.00 (d, 1H,  $J$  1.8 Hz, C<sup>3</sup>H), 3.82 (dd, 1H,  $J$  7.8 and 1.8 Hz, C<sup>4</sup>H), 1.31 [s, 9H, C(CH<sub>3</sub>)<sub>3</sub>], 1.21 [s, 9H, C(CH<sub>3</sub>)<sub>3</sub>], 0.78–0.61 (m, 4H, SiCH<sub>2</sub>CH<sub>2</sub>Si), 0.12 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>], 0.09 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>]. (**4b**):  $\delta$  7.53 (d, 1H,  $J$  7.6 Hz, HC=N), 7.41–7.13 (m, 10H, arom.), 4.85 [q, 1H,  $J$  7.2 Hz, HC(CH<sub>3</sub>)Ph], 4.34 [q, 1H,  $J$  6.6 Hz, HC(CH<sub>3</sub>)Ph], 4.18 (d, 1H,  $J$  1.9 Hz, C<sup>3</sup>H), 3.67 (dd, 1H,  $J$  7.6 and 1.9 Hz, C<sup>4</sup>H), 1.50 [d, 3H,  $J$  6.6 Hz, CH<sub>3</sub>C(H)Ph], 1.39 [d, 3H,  $J$  7.2 Hz, CH<sub>3</sub>C(H)Ph], 0.76–0.52 (m, 4H, SiCH<sub>2</sub>CH<sub>2</sub>Si), 0.05 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>], –0.05 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>]. (**4c**):  $\delta$  8.62–8.59 (m, 1H, pyr.), 7.67–7.58 (m, 1H, pyr.), 7.32–7.09 (m, 7H, arom. and pyr.), 5.01 [q, 1H,  $J$  7.2 Hz, CH(CH<sub>3</sub>)Ph], 4.36 (d, 1H,  $J$  2.0 Hz, C<sup>3</sup>H), 4.04 (d, 1H,  $J$  2.0 Hz, C<sup>4</sup>H), 1.26 [d, 3H,  $J$  7.2 Hz, CCH<sub>3</sub>(H)Ph], 0.70–0.56 (m, 4H, SiCH<sub>2</sub>CH<sub>2</sub>Si), –0.09 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>], –0.13 [s, 6H, Si(CH<sub>3</sub>)<sub>2</sub>].

‡ This is an uncorrected enantiomeric excess (e.e.); the optical purity of the starting (*R*)- $\alpha$ -methylbenzylamine was 93%. Therefore the optical yield of (**4b**) is better than 90%.